Platinum Group Metal Chelates Derived from 2-Mercapto-, 2-Hydroxy- and 2-Amino-pyridines and -pyrimidines

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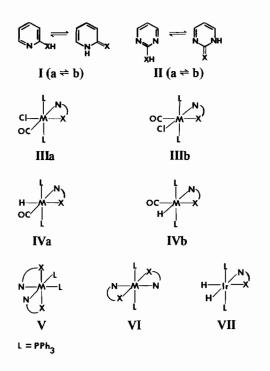
The title ligands I and II (X = S, O or NH) are amongst the simplest molecules containing the tautomeric system:

$$-NH-C=X \implies -N=C-XH$$

and are therefore useful model compounds for a range of important biological molecules – notably purine and pyrimidine bases – which encompass the same groupings. In view of current interest in the bonding of DNA bases to platinum group metals [1, 2], we have begun to investigate the reactions of the title ligands, I and II (X = S, O or NH), with a range of platinum group metal complexes. In this preliminary report we describe products obtained using low oxidation state or hydrido complexes of the rarer platinum metals.

A representative selection of precursors used together with products obtained from 2-mercapto-, 2-hydroxy-, and 2-amino-pyridines is presented in Table I, reaction conditions employed are given in footnotes to the same. All the complexes isolated are air-stable crystalline solids ranging in colour from

TABLE I. Precursors and Products Obtained, L = PPh3.



white through pale yellow to orange. Stereochemical assignments have been made where possible using <sup>1</sup>H, <sup>13</sup>C and <sup>31</sup>P NMR data.

A few of the complexes listed in Table I have been previously reported [3, 4]. However the present work affords the first general method for the preparation of this class of chelates.

The complexes MCl(pyX)(CO)(PPh<sub>3</sub>)<sub>2</sub> (M = Ru or Os, X = O or S) all have *trans*-phosphine ligands  $[^{31}P{^{1}H}$  NMR singlets] and thus possess stereochemistry III (a or b). However, the osmium complex OsCl(pyO)(CO)(PPh<sub>3</sub>)<sub>2</sub> has also been observed

Precursors	pySH Products	pyOH Products	pyNH <sub>2</sub> Products
RuHCl(CO)L <sub>3</sub>	$RuCl(pyS)(CO)L_2^{a}$	RuCl(pyO)(CO)L <sub>2</sub> <sup>b</sup> RuH(pyO)(CO)L <sub>2</sub> <sup>b</sup>	
$RuH_2(CO)L_3$	RuCl(pyS)(CO)L <sub>2</sub> <sup>a</sup> RuH(pyS)(CO)L <sub>2</sub> <sup>b</sup>	RuH(pyO)(CO)L <sub>2</sub> <sup>b</sup>	RuH(pyNH)(CO)L <sub>2</sub> <sup>a</sup>
$Ru(CO)_3L_2$	Ru(pyS) <sub>2</sub> (CO)L <sup>b</sup>	RuH(PyO)(CO)L <sub>2</sub> <sup>b</sup>	
RuH <sub>2</sub> L <sub>4</sub>	$Ru(pyS)_2L_2^a$	$Ru(pyO)_2L_2^b$	
RuCl <sub>2</sub> L <sub>3</sub>		$RuCl_2(pyO)L_2^c$	
RuCl <sub>2</sub> L <sub>3</sub>	Ru(pyS) <sub>2</sub> L <sub>2</sub> <sup>d</sup>	$Ru(pyO)_2L_2^e$	$Ru(pyNH)_2L_2^d$
$RuCl_2(CO)_2L_2$		RuCl(pyO)(CO)L <sub>2</sub> <sup>e</sup>	RuCl(pyNH)(CO)L <sub>2</sub>
OsHC1(CO)L <sub>3</sub>	OsCl(pvS)(CO)L <sub>2</sub> <sup>b</sup>	OsCl(pyO)(CO)L <sub>2</sub> <sup>b</sup>	
OsH <sub>2</sub> (CO)L <sub>3</sub>	OsCl(pyS)(CO)L <sub>2</sub> <sup>b</sup> OsH(pyS)(CO)L <sub>2</sub> <sup>b</sup>	OsCl(pyO)(CO)L <sub>2</sub> <sup>b</sup> OsH(pyO)(CO)L <sub>2</sub> <sup>f</sup>	
OsH <sub>4</sub> L <sub>3</sub>	$Os(pyS)_2L_2^a$		
IrH <sub>3</sub> L <sub>3</sub>	IrH <sub>2</sub> (pyS)L <sub>2</sub> <sup>a</sup>	$IrH_2(pyO)L_2^{f}$	

<sup>a</sup>Refluxing benzene. <sup>b</sup>Refluxing toluene. <sup>c</sup>Warm benzene/O<sub>2</sub>. <sup>d</sup>Refluxing benzene/NEt<sub>3</sub>. <sup>e</sup>Refluxing toluene/NEt<sub>3</sub>. <sup>f</sup>Refluxing 2-methoxyethanol.

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as a kinetically controlled cis-phosphine isomer  $[^{31}P{^{1}H}$  NMR AB pattern ( $\delta$  5.90 and 3.40 ppm <sup>2</sup>J(PH) 12 Hz]. The hydrides MH(pyX)(CO)(PPh<sub>3</sub>)<sub>2</sub> exist as a mixture of two isomeric forms IV (a and b) when X = O or  $S [^{1}H NMR, 2$  high field triplets, <sup>31</sup>P{<sup>1</sup>H}NMR two singlets] but afford only a single isomer IV (a or b) when X = NH. The bis-chelates  $M(pyX)_2(PPh_3)_2$  (X = O, S or NH) display a fine balance between cis- and trans-phosphine structures (V and VI). The complexes Ru(pyS)<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> [3] and  $Ru(6-Me-pyO)_2(PPh_3)_2$  [4] have previously been shown by X-ray diffraction methods to adopt cis- and trans-phosphine structures respectively. However NMR data clearly establish that the ruthenium complexes  $Ru(pyX)_2(PPh_3)_2$  and their osmium analogs adopt trans stereochemistry in solution (C6- $H_6$  or CHCl<sub>3</sub>). Thus in each case the ortho to meta/ *para* proton separation for solutions in  $C_6D_6$  is greater than 0.5 ppm indicating a trans triphenylphosphine arrangement [5]. This stereochemical assignment is confirmed for the complexes M(pyS)2- $(PPh_3)_2$  by the  ${}^{13}C{}^{1}H$  NMR spectra which con-tain  ${}^{13}C{}^{-31}P$  virtual coupling triplet patterns for all triphenylphosphine carbon atom save those in the para position [6] but display no evidence of <sup>13</sup>C-<sup>31</sup>P couplings in the resonances of the pyS carbon atoms. The latter result is particularly significant since, in our experience, structures with pyS trans to PPh<sub>3</sub> ligands do show such couplings.

The ruthenium(III) products  $RuCl_2(pyX)(PPh_3)_2$ , obtained from  $RuCl_2(PPh_3)_3$  and pyXH in the presence of air, are analogs of the previously reported osmium complex  $OsCl_2(pyO)(PPh_3)_2$  [7] and are closely related to the known ruthenium(III) and osmium(III) carboxylates  $MCl_2(O_2CR)(PPh_3)_2$  [8]. Finally the iridium dihydrides  $IrH_2(pyX)(PPh_3)_2$ each display NMR resonances [<sup>1</sup>H, high field 2 doublets of triplets; <sup>31</sup>P{<sup>1</sup>H}, singlets.] indicative of stereochemistry (VII). This structure has recently been confirmed for  $IrH_2(pyS)(PPh_3)_2$  by X-ray crystallography [9].

Analogous complexes containing chelate ligands derived from 2-mercapto-, 2-hydroxy-, and 2-aminopyrimidines are in the course of preparation and attempts to extend the preparative method to include the DNA bases – adenine, cytosine, guanine, and thymine – are in progress.

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